STERIC EFFECTS ON THE CONFORMATION OF VALEROLACTORES

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Apart from incidental observations, the conformation of δ -lactones has, as far as we are aware, been the subject of three generalisations (1,2,3). Although Mathieson (1) inferred generally the boat conformation, more recently "normal" carbonyl-stretching frequencies near 1740 cm⁻¹ in solution have been correlated with half-chair conformations in the solid state by X-ray analysis, and "high" values near 1760 cm⁻¹ in solution have been correlated with boat conformations in the solid state by X-ray analysis, or geometrical requirements in bridged systems (2). Most recently optical rotatory dispersion measurements on optically active δ -lactones derived from natural products of known configuration have been correlated on the basis that the δ -lactone ring adopts a boat conformation unless certain steric interactions are relieved in a half-chair conformation (3).

We have examined the proton magnetic resonance (p.m.r.) spectra of all twelve isomeric gem-dimethylphenylvalerolactones and of 3,3-dimethyl-5-tert-butylvalerolactone. Vicinal coupling constants or pair sums readily extracted from the spectra by appropriate analyses are shown in Table 1, where for consistency, we refer to the 2-, 3-, 4-, and 5-methylene groups as AB, KL, ME, and XX, respectively, A, K, M, and X being the downfield

Table 1

COUPLING CONSTANTS FOR GEN-DIMENTYLPHRAYLVALEROLAGTONES

in CDC1, at 30°

	Substituents		Coupling constants (c/s)			<pre># (J0 in c/s)</pre>
	No ₂	Ph.	Vicinal		Long-renge	
I	2,2	4	<u>MX</u> 5.0	MT 11.0		ø ₄₅ 49° (11.6)
			KM 12.8	<u>IM</u> 3.5	<u>IX</u> 2.2	ø ₃₄ 58° (12.5)
n	5,5	3	<u>₩</u> 4.3	<u></u> 12.7		ø ₂₃ 55° (13.1)
			<u>KM</u> 3-5	EE 12.6	A¥ 2.0	ø ₃₄ 58° (12.5)
ш	3,3	5	<u>XX</u> 3.1	IX 12.7	AX 1.7	ø ₄₅ 60° (12.4)
IV	4,4	2	AX 6.5	<u>AL</u> 13.1	1.5	ø ₂₃ 46° (13.5)
¥	3,3	4	<u>mx</u> 10.9	MT 5.0		Δ <u>σ</u> °>1760
VI	4,4		<u>₩</u> 11.5 ⁸			
AII	2,2	3	$(\overline{n}\overline{x} + \overline{n}\overline{x})_{\overline{p}}$			
			(MI + MI)	8.7		<u>∆ g</u> °∼680
			KM 11.2	K# 3.0		
AIII	3,3	2	<u>m</u> 5.6	<u>nx</u> 9.0		
			<u>NY</u> 5.5	EX 5•4		<u>∆</u> <u>g</u> °
IX	4,4	5	(W + K)	16.0		
			(VT + RT) ₆	13.1		<u>∆</u> <u>g</u> ° ~ 240

Approx. values to fit calculated ABC spectrum to observed 8-line spectrum

or only members of the respective pairs. Lactones I-IV each have the phenyl group in a cis-1,3 relation to one of the geminal methyl pair, and may be expected to exist preferentially in a conformation that avoids diaxial opposition of these groups and probably eclipsing interaction of especially the gem-dimethyl pair with an adjacent methylene group. First approximations, probably a little high (cf. 4), to violate cis-dihedral

 $[\]frac{\mathbf{b}}{\mathbf{x}}$ and $\underline{\mathbf{x}}$ assignments are arbitrary

C K and L assignments are arbitrary

angles as given by solution of

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are consistent with existence of these four in fixed half-chair conformations in deuterochloroform at 30°, the phenyl group being equatorial (or pseudo-equatorial at the 2- or 5- positions). More diagnostic is the occurrence for all four of marked long-range coupling between the more weakly vicinally coupled (equatorial) proton of an appropriate methylene group and one of the protons of the other methylene group, and of unresolved long-range coupling (detected by spin-decoupling) of one of the methyl groups and the adjacent more strongly vicinally coupled (axial) methylene proton. The near-planar W-arrangement of bonds usually required for such long-range coupling (5) is present in the half-chair but not in the boat conformation. The efficacy of the phenyl-methyl interaction in fixing conformation in these cases is attested by the values, $\underline{J}_{\overline{M}} = 3.0$, $\underline{J}_{\overline{M}} = 12.5$, $\underline{J}_{\overline{M}} = 1.7$ o/s for 3,3-dimethyl-5-tert-butylvalerolactone (cf. III, Table 1).

All other lactones in Table 1 have adjacent phenyl and gem-dimethyl substituents which are unlikely to be eclipsed in a stable conformation. This implies half-chair conformations for V and VI. All thirteen valerolactones reported on here show in carbon tetrachloride solution "normal" carbonyl frequencies in the infrared (1730-1740 cm⁻¹ if the gem-dimethyl group is 2- or 5-, 1740-1750 cm⁻¹ if it is 3- or 4-), so there is probably no significant incursion of boat forms. The p.m.r. spectrum of III is little affected by increase of temperature in the range, 30-150°, but that of V shows marked changes probably as a result of increasing residence times in a less stable inverted conformation. Vicinal coupling constants in comparison with those for I suggest the phenyl group in V is >95% equatorial in CDCl₂ at 30° (whence $\Delta \underline{g}^{0}$ > 1760 cal/mole). The ABC spectrum of VI is more difficult to analyse but a similar conclusion probably applies. Lectones VII-IX each contains a vicinal pair of methylene groups and application of the approximation previously suggested (6) for rapid symmetrical inversion gives the $\Delta \underline{c}^{0}$ values shown. If, as is clearly so

for lactones I-VII, the equatorial-phenyl conformer is the more stable, the $\Delta \underline{\mathbf{c}}^0$ values imply that interaction of a <u>pseudo-axial</u> phenyl group at the 2- or 5- positions with an axial 4- or 3- hydrogen atom, respectively, and the lone pair or pi-electrons of the lactone group is less energetic than the corresponding interactions of a more nearly axial phenyl group at the 3- or 4- positions.

The remaining isomeric gen-dimethylphenylvalerolactones show closely coupled 5-spin spectra we have not yet analyzed unambiguously.

Synthesis and characterisation of new compounds will be reported elsewhere. P.m.r. spectra were measured on a Varian HA-60-IL spectrometer purchased with a grant from the Australian Research Grants Committee.

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